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 Ani-Mpei Cheu-Yaug
 041781-2016
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 VPPLICATION NO:
 FILING DATE
 FIRST NAMED INVENTOR
 ATTORNEY DOCKET NO:
 CONFIRMATION NO:

ART UNIT APPER NUMBER APPER 1246

96/1

DATE MAILED: 12/16/2003

Please find below and/or attached an Office communication concerning this application or proceeding.

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11) The oath or declaration is objected to by the	Examiner. Note the attache	dflice Action or form PTO-152.	
Replacement drawing sheet(s) including the cor			
Applicant may not request that any objection to	he drawing(s) be held in abeya	ice. See 37 CFR 1.85(a).	
10) The drawing(s) filed on is/are: a) □	ccepted or b)□ objected to	by the Examiner.	
9) The specification is objected to by the Exam	ner.		
Papers Papers			
8) Claim(s) are subject to restriction an	d/or election requirement.		
7) \boxtimes Claim(s) 12 and 13 is/are objected to.			
Claim(s) $\frac{1-11 \text{ and } 14}{1}$ is/are rejected.			
5) Claim(s) is/are allowed.			
4a) Of the above claim(s)			
4) \boxtimes Claim(s) $\frac{1-14}{1-14}$ is/are pending in the applicate	·uo		
sposition of Claims			
3) Since this application is in condition for allo closed in accordance with the practice und			
Sa)☐ This action is FINAL . Sb)☑ T	is action is non-final.		
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after SIX (6) MONTHS from the mailing date of this communication - If the pendod for reply specified above, the maximum statutory pen - If MO pendod for reply is specified above, the maximum statutory pen - Failure to reply within the set or extended pendod for reply will, by st - Any reply received by the Office later than three months after the m earned patent term adjustment. See 37 CFR 1.704(b).	od will spply and will expire SIX (6) MO tute, cause the application to become A	THS from the mailing date of this communication (35 U.S.C. § 133).	
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Application/Control Number: 09/865,478

Art Unit: 1746

DETAILED ACTION

Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 1-4, 7-11 & 14 are rejected under 35 U.S.C. 102(b) as being anticipated by Coltrain et al. U.S. Patent 5,01028.

Coltrain teaches a solid composite polymer electrolyte comprising a branched polymer of poly[bis(methoxyethoxyethoxy)phosphazene] or (MEEP) (Preparation 2) having a molecular weight of 1x10⁴ to 1 x 10⁶ (par. 24, claim 4). The MEEP inherently is a general amorphous branched polymer having recurrent units, each of which includes a backbone chain and at least a side chain linked to said backbone chain and containing at least one coordination potential atom; wherein the branched polymer is a -P=N - group, and said coordination potential atom is an alkoxy group (claims 2-3). The solid composite includes Al₂O₃ and TiO₂ (par. 40, claim 7) which are inherently amphoteric Lewis acid-base ceramic fillers that form Lewis acid-base interactions with side chains of the branched polymer and an amphoteric metal salt. The ceramic filler has a particle size of less than 0.2 microns (par. 36, claim 14). Metallic salts are added to the polymer electrolyte including lithium salt (claim 8) such as LiClO₄ (par. 34, claim

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9). The salt is inherently amphoteric and forms Lewis acid-base interaction with side chains of the branched polymer. The MEEP may be in an amount of 20% to about 95% by weight (par. 39), the ceramic filler is found in an amount of 1 to 80% (par. 40) and the salt is between about 5 to about 20% by weight (par. 41) (claim 10-11). Therefore, the instant claims are anticipated by Coltrain.

Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States

Claims 1,2,5,6,7,8 are rejected under 35 U.S.C. 102(b) as being anticipated by Kumar U.S. Patent 5,695,873.

Kumar teaches a solid composite polymer electrolyte comprising polyacrilonitrile(PAN) (col. 2, lines 10-20). The PAN inherently is a general amorphous branched polymer having recurrent units, each of which includes a backbone chain and at least a side chain linked to said backbone chain and containing at least one coordination potential atom; wherein the branched polymer is a −C-C − group, and said coordination potential atom is an C=N group (claims 2 & 5). The PAN inherently has a molecular weight within the range from about 10000 to about 1 x 10⁷ (claim 6). The

solid composite includes Al₂O₃ and (col. 2, lines 10-20, claim 7) which are inherently amphoteric Lewis acid-base ceramic fillers that form Lewis acid-base interactions with side chains of the branched polymer and an amphoteric metal salt. A metallic salt is added to the polymer electrolyte including lithium salt of LiBF₄ (par. 34, claim 8). The salt is inherently amphoteric and forms Lewis acid-base interaction with side chains of the branched polymer.

Allowable Subject Matter

Claim12 and 13 are objected to as being dependent upon a rejected base claim, but would be allowable if rewritten in independent form including all of the limitations of the base claim and any intervening claims.

Claim 12 would be allowable over the prior art of record, because the prior art is silent to a solid composite polymer comprising 41 to 70% by weight PAN, 27 to 50% of lithium perchlorate and 3 to 9 % of ceramic filler.

Claim 13 would be allowable over the prior art of record, because the prior art is silent to a solid composite polymer comprising 47 to 60% by weight PAN, 35 to 45% of lithium perchlorate and 5 to 8 % of ceramic filler.

The prior art, such as Kumar et al., teaches a polymer ceramic composite electrolyte wherein 25 to 60% by weight of the electrolyte contains a ceramic filler. Further the lithium salt is present in an amount of 10 to 20%. The reference is silent to

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a ceramic filler content of less than 9% necessitated by the instant claims. The crux of Kumar's invention includes a ceramic filler of at least 25% to enhance cationic transport. Therefore, the instant claims are patentably distinct from Kumar.

Conclusions

Any inquiry concerning this communication or earlier communications from the Examiner should be directed to Monique Wills whose telephone number is (703) 305-0073. The Examiner can normally be reached on Monday-Friday from 8:30am to 5:00 pm.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the Group receptionist whose telephone number is (703) 308-0661.

If attempts to reach Examiner by telephone are unsuccessful, the Examiner's supervisor, Randy Gulakowski, may be reached at 703-308-4333.

The unofficial fax number is (703) 305-3599. The Official fax number for non-final amendments is 703-872-9310. The Official fax number for after final amendments is 703-872-9311.

Mw

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